

Determination of lanthanum...

S/032/62/028/006/013/025  
B101/B138

$\text{Ce}_2\text{O}_3$  does not affect Y determination. A laboratory assistant is able to analyze 10 - 15 samples three times each during one shift.

ASSOCIATION: Kirgizskiy gosudarstvennyy universitet (Kirgiz State University)

Card 2/2

BELYAYEV, V.P.; KALINACHENKO, V.R.; KUZ'KIN, N.M.; YAKIMENKO, L.M.;  
ARGYANULAEV, V.N.; RUBENZHIK, Yu.I.; SLEVKUN, I.G.;  
SHKLOVER, L.P.; BURAVLEV, Yu.M.; PEREPELKINA, M.A.;  
USTINNOVA, V.I.; NEUYNINA, G.P.; ENGEL'SHT, V.S.; TRAPITSYN, N.F.;  
BULANOV, Yu.A.

Exchange of experience. Zav.lab. 28 no.6:685-687 '62.

(MIRA 15:5)

1. Khimicheskiy zavod imeni Vaykova (for Shklover). 2.  
Ural'skiy nauchno-issledovatel'skiy institut chernykh metallov  
(for Buravlev, Perepelkina, Ustinova, Neuymina). 3. Kirgizskiy  
gosudarstvennyy universitet (for Engel'sht, Trapitsyn, Bulanov).  
(Spectrum analysis)

ZAYCHIK, Isay Tur'yevich, inzh., USOV, Sergey Nikolayevich, inzh.;  
CHISTYAKOV N.I., doktor tekhn. nauk, prof., retsenzent;  
BULANOV, Yu.A., prepredavatel', inzh., retsenzent; BRAMMER,  
Yu.A., kand. tekhn. nauk, nauchn. red.; BASAVINA, Ye.V.,  
red.

[Textbook on amplifying and radio receiving devices] Za-  
dachnik po usilitel'nym i radiopriemnym ustroistvam. Mc-  
skva, Vysshaisa shkola, 1965. 315 p. (MIRA 18:11)

1. Moskovskiy elekrotekhnicheskiy institut svyazi (for  
Chistyakov). 2. Moskovskiy tekhnikum avtomatiki i telemek-  
haniki (for Bulanov).

ACCESSION NR: AP4042620

S/0096/64/000/008/0054/0057

AUTHOR: Gulyayev, V. N. (Candidate of technical sciences); Tseytlin, V. Z. (Candidate of technical sciences); Ryabova, L. I. (Engineer); Talov, N. P. (Engineer); Bulanov, Yu. P. (Engineer)

TITLE: Effect of the duration of heating on the structure and properties of chromium-manganese-nickel steels

SOURCE: Teploenergetika, no. 8, 1964, 54-57

TOPIC TAGS: chromium manganese nickel steel, austenitic heat resistant steel, low nickel steel, austenitic steel, steel aging, steel corrosion, austenitic steel steam pipeline, OKh14N3G11AB steel, OKh18N5G12AB steel, 1Kh14N3G14T steel, 1Kh18N9T steel

ABSTRACT: In a search for substitutes for 1Kh18N9T (AISI321) steel in high-temperature steam service, the structure, phase composition, mechanical properties, and susceptibility to intergranular corrosion of three heat-resistant, stainless, low-nickel steels have been investigated after aging at 500, 550, and 650C for up to 2000 hr. Induction-melted ingots of the OKh14N3G11AB steel, OKh18N5G12AB steel,

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and 1Kh14N3G14T steel were forged and air cooled from 1050C. In the 20—650C temperature range, the strength of the new steels in the initial state was equal to or higher than that of 1Kh18N9T steel. The room-temperature ductility of all the steels except OKh18N5G12AB was higher than that of 1Kh18N9T steel. At room temperature, OKh14N3G11AB steel had a notch toughness of 14—19 kgm/cm<sup>2</sup>, OKh18N5G12AB steel, of 7—13 kgm/cm<sup>2</sup>, and 1Kh14N3G14T steel, of 26—32 kgm/cm<sup>2</sup>. Aging of Cr-Mn-Ni steels at 500C or higher produced diffusional decomposition of the supersaturated solid-solution austenite with the precipitation of chromium and manganese carbides and nitrides, predominantly along the grain boundaries. The diffusional decomposition of austenite of nitrogen-containing Cr-Mn-Ni steels induces hot brittleness in them, particularly in OKh18N5G12AB steel, whose notch toughness dropped to 2—4 kgm/cm<sup>2</sup> after 2000-hr aging at 650C. The steels became susceptible to intergranular corrosion after about 100-hr aging at 500C; however, the corrosion resistance gradually increased after about 1000-hr aging. In general, the investigated steels should not be used at temperatures higher than 460—470C when the operating conditions might promote intergranular corrosion by water and/or steam. In the absence of such a medium, an operating

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ACCESSION NR: AP4042620

temperature as high as 500C can be permitted, with no changes occurring in the structure or mechanical properties. Orig. art. has: 6 figures and 2 tables.

ASSOCIATION: VTI; TsNIIChM

SUBMITTED: 00

ATD PRESS: 3083

ENCL: 00

SUB CODE: MM

NO REF Sov: 004

OTHER: 000

Card 3/3

ACCESSION NR: AP4040987

S/0279/64/000/003/0145/0147

AUTHORS: Gulyayev, V. N. (Chelyabinsk); Bulanov, Yu. P. (Chelyabinsk)

TITLE: Phases of  $Ti_nNi_mC$  in steels 1Kh18N12T and 1Kh18N9T

SOURCE: AN SSSR. Izvestiya. Metallurgiya i gornoye delo, no. 3, 1964, 145-147

TOPIC TAGS: steel, titanium, nickel, carbon, grain structure, phase property, sigma phase/ R&amp;D x ray camera, 1Kh18N12T steel, 1Kh18N9T steel

ABSTRACT: With the aim of determining the causes of damage of straight tubes made of steels 1Kh18N12T and 1Kh18N9T and of overcoming the discrepancies encountered in the behavior of tubes with fine grain structure, the authors conducted a study of the phase properties of the tube metals after keeping them for 10 000 to 15 000 hours at a temperature between 600 and 615°C. The phase properties were studied by means of electrochemical separation of the phases and subsequent chemical and radiographic analyses. For the differentiation of the phase components, anode dissolution was used in two component electrolytes. For strong acids 250 ml HCl + 150 ml H<sub>2</sub>O + 5 g oxalic acid was used at a surface current of 0.03-0.05 amp/cm<sup>2</sup>. For weak acids, 200 ml HCl + 1000 ml H<sub>2</sub>O + 5 g oxalic acid or 350 g KCl + 950 ml

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ACCESSION NR: AP4040987

$H_2O + 50 \text{ ml HCl}$  was used at  $1 \text{ amp/cm}^2$ . The radiographic analysis was performed by the x-ray method using an RKD camera. The results of the dependence of the residual deformation due to creep on the appearing phases showed the role of  $Ti_nNi_mC$ . For a residual deformation of 0.1-1.56%, the phases observed were TiC,  $Me_{23}C_6$  and the sigma phase. For a residual deformation of 2.5-3.13%,  $Ti_nNi_mC$  was also observed. For 2.5-12.56%, only TiC, sigma, and  $Ti_nNi_mC$  phases were present. The authors thank the scientific collaborator comrade L. N. Rastorguyev of Moskovskiy institut stali i splavov (Moscow Institute of Steel and Alloys) for his help in carrying out this work. Orig. art. has: 1 figure and 1 table.

ASSOCIATION: none

SUBMITTED: 20May63

SUB CODE: MM

NO REF SOV: 007

ENCL: 00

OTHER: 001

Card 2/2

L15735-65 ENT(m)/EMP(w)/EWA(d)/EMP(t)/EMP(r)/EMP(b)  
MJW/JD/HW

Pf-4 ASD(m)-3

ACCESSION NR: AP4047992

S/0096/64/000/011/0068/0071

AUTHORS: Gulyayev, V. N. (Candidate of technical sciences); Bulanov, Yu. P.  
(Engineer)

B

TITLE: Failure of steam superheat pipes made from steels Kh18N12T and 1Kh18N9T

SOURCE: Teploenergetika, no. 11, 1964, 68-71

18

18

TOPIC TAGS: steel pipe failure, nickel steel, titanium steel, steam pipe/1Kh18N12T  
steel, 1Kh18N9T steel, NI 257 steel

18

ABSTRACT: In order to determine the cause (or causes) of a number of failures in steam superheat pipes and elbows (made of Kh18N12T and 1Kh18N9T steels) at an operating temperature of 610C, 23 damaged and undamaged pipes (32 mm diameter, 5.5 mm wall) were investigated after 11 126-15 505 hours of operation. The physical properties (impact strength, elongation, tensile strength) as well as grain size and contents of different phases were investigated with the following conclusions:  
1) the properties of the pipe materials, which satisfy the requirements of technical specifications ChMTU 2884-51, sometimes do not correspond to the properties of the material which was used in the experiments on which the original recommendation of steel 1Kh18N12T (1Kh18N9T) for this application was based; 2) failure can occur

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L 15735-65  
ACCESSION NR: AP4047992

even though the requirements of technical specifications ChMTC-2884-51 are satisfied and even though the operating pressures and temperatures do not exceed the design values. A major reason for these failures is the separation of the  $Ti_{n,m}NiC$  phase;

3) structural instability in pipes made from the above steels is caused by small grain size (7, 8 units and smaller on the grain size scale); 4) pipes made of 1Kh18N12T steel offer no advantages over pipes made of 1Kh18N9T and thus represent a useless waste of valuable Ni; 5) during repairs, pipes made of 1Kh18N9T should be used after inspection of grain size (4-6 units), and should have carbon and titanium content just sufficient to produce TiC; 6) in view of the detrimental effect of  $Ti_{n,m}NiC$ , the necessity of alloying with titanium should be reviewed. Orig. art. has: 3 figures.

ASSOCIATION: VoF VTI

SUBMITTED: 00

ENCL: 00

SUB CODE: MM, IE

NO REF SOV: 007

OTHER: 001

Card 2/2

L 10302-67 EWT(m)/EWP(t)/ETI/EWP(k) IJP(c) JD/iW  
ACC NR: AP7003103

SOURCE CODE: UR/0096/66/000/006/0042/0043

26

AUTHOR: Krutasova, Ye. I. (Candidate of technical sciences); Venkova, L. F.  
(Engineer); Bulanov, Yu. P. (Engineer)

ORG: none

TITLE: Structure and properties of tube metal made of 12Kh2MFSSR steel after long  
aging

SOURCE: Teploenergetika, no. 6, 1966, 42-43

TOPIC TAGS: metal tube, steel microstructure

ABSTRACT: Results are presented from an investigation of the change in properties of 12Kh2MFSSR tube steel after long aging at high temperatures under laboratory and usage conditions in units with ultrahigh steam parameters. Photographs of the microstructure of the metal are presented. Investigation showed that after aging up to 10,000 hr at 620°C, a new phase is separated at the boundaries of the ferrite grains which apparently increases the resistance of the steel to the action of high temperatures. At 620°C in an atmosphere of air and fuel gases, scale formation proceeds rapidly on this steel. Orig. art. has: 5 figures and 2 tables.  
[JPRS: 37,415]

SUB CODE: 13, 11 / SUBM DATE: none

Card 1/1

UDC: 621.772.4.620.183.001.45

BULANOVA, A. I.

USSR/Physics - Coercive Force

Aug 52

"Problem of Determining the Coercive Force of Non-uniformly Magnetized Samples," A. I. Bulanova, M. N. Mikhnevich, V. B. Perets.

"Zhur Tekh Fiz" Vol 22, No 8, pp 1325-1333

Ballistic methods were applied to measurements of coercive force of nonuniformly magnetized cylindrical samples. Compares exptl mental and theoretical results. States that choice of the correct method of measurement is dictated by conditions and purpose. Indebted to Prof R. I. Yanus. Received 28 Jun 50.

226T99

Translation from: Referativnyy Zhurnal, Elektrotehnika, 1957,  
Nr 3, p. 158 (USSR) 112-3-6146

AUTHOR:

Bulanova, A.I., Veksler, A.Z., Rudnyy, N.M.

TITLE:

Investigation of the Wattmeter Method of Measuring Losses  
in Simultaneous Magnetization of Electric Steel by Static  
and Dynamic Fields (Issledovaniye vattmetrovogo metoda  
izmereniya poter'pri odnovremennom namagnichivaniil elektro-  
tekhnicheskoy stali postoyannym i peremennym polyami)

PERIODICAL: Tr. Vses. n.-i. in-ta metrol., 1956, Nr 29 (89), pp. 127-  
138 in

ABSTRACT:

By using the wattmeter method/investigating installations  
for determining losses in double magnetization, using  
individual feed circuits for the sample under test and a  
common winding for direct and alternating currents, it  
was established that the common winding gave the smallest  
errors in measuring losses. The variable component of  
field intensity is measured by a special electrodynamic  
ammeter with a compensating winding, through which passes

Card 1/2

BULANOV, A.I.

24(0); 5(4); 6(2) PHASE I BOOK EXPLOITATION SOV/2215

Vsesoyuznyy nauchno-issledovatel'skiy institut metrologii imeni D.I. Mendeleyeva  
Referaty nauchno-issledovatel'skiy rabot; sbornik No. 2 (Scientific Research Abstracts; Collection of Articles, Nr 2) Moscow, Standardgiz, 1958. 139 p. 1,000 copies printed.

Additional Sponsoring Agency: USSR. Komitet standartov, mer i izmeritel'nykh priborov.

ED.: S. V. Reshetina; Tech. Ed.: M. A. Kondrat'yeva.

PURPOSE: These reports are intended for scientists, researchers, and engineers engaged in developing standards, measures, and bases for the various industries.

COVERAGE: The volume contains 123 reports on standards of measurement and control. The reports were prepared by scientists of the Institute of the Komitet standartov, mer i izmeritel'nykh priborov pri Sovete Ministerov SSSR (Commission on Standards, Measures, and Measuring Instruments under the USSR Council of Ministers). The participating institutes are: VNIM, Vsesoyuznyy nauchno-issledovatel'skiy metrologicheskii institut D.I. Mendeleyeva (All-Union Scientific Research Institute of Metrology imeni D.I. Mendeleyeva) in Leninsk; Sverdlovsk branch of this institute; VNIIK - Vsesoyuznyy nauchno-issledovatel'skiy Institut Komiteata standartov, mer i izmeritel'nykh priborov (All-Union Scientific Research Institute of the Commission on Standards, Measures, and Measuring Instruments), created from MGIMIP - Moskovskiy Gosudarstvennyy Institut mer i izmeritel'nykh priborov (Moscow State Institute of Measures and Measuring Instruments) October 1, 1955; VNIPIPII - Vsesoyuznyy nauchno-issledovatel'skiy institut chislovo-tekhnicheskikh radioelektronnykh sistem (All-Union Scientific Research Institute of Radioelectronics, Computers, and Radio-engineering Measurements) in Novosibirsk; KhGIAM - Khar'kovskiy Gosudarstvennyy Institut mer i izmeritel'nykh priborov (Kharkov State Institute of Measures and Measuring Instruments); and NGIMIP - Novosibirskiy Gosudarstvennyy Institut mer i izmeritel'nykh priborov (Novosibirsk State Institute of Measures and Measuring Instruments); no personnel are mentioned. There are no references.

RUDYK, N.M., and A.I. BULANOV (Sverdlovsk Branch of VNIM). Diving Lenses Between Asphericals and Eddy Currents in Electrical Steel 105

RUDYK, N.M., A.I. BULANOV, and A.Z. Vekseler (Sverdlovsk Branch of VNIM). Smelzhenie zerkal i oshibki v usloviyakh nejednorodnosti. (Smelzhenie zerkal i oshibki v usloviyakh nejednorodnosti) 106

STRAKUN, G.I. (VNIM). Requirements of Optical Systems Used to Making a System Satisfying These Requirements 109

STRAKUN, G.I. (VNIM). Studying Lenses for Checking Dioptr 107

STRAKUN, G.I. (VNIM). Effect of Aberrations of Objective Lenses Used to Photograph Interference Patterns on the Distribution of Illumination over the Images of Rings of Equal Inclination 107

STRAKUN, G.I. (VNIM). Requirements of Optical Systems Used to Making a System Satisfying These Requirements 109

BULANOVA, A.I.

Scientific Research Abstracts; (Cont.)	SOV/2215
and Voltmeters at High Frequencies	97
<u>Bezikovich, A.Ya.</u> (VNIIM). Errors of Electrodynamic Wattmeters at High Frequencies	100
<u>Lubentsov, V.F.</u> , <u>S.M. Okhotina</u> , and <u>P.A. Shpan'on</u> (KhGIMIP). Apparatus for Checking Tube Voltmeters	101
<u>Rumyantsev, A.S.</u> , and <u>Ye. P. Dubovik</u> (VNIIM), and <u>A.A. Chukh-</u> <u>lantsev</u> (Sverdlovsk Branch of VNIIM). Developing Methods and Standard Apparatus for Testing Direct-Current Transformers Type I-58 Under Operating Conditions at 70 Kiloamperes	102
<u>Lizogub, M.S.</u> , <u>V.I. Zingerman</u> , and <u>Ye. Ye. Bogatyrev</u> (KhGIMIP). Developing and Studying Apparatus for Measuring Magnetic Fields by the Nuclear Magnetic Resonance Method	102
<u>Rudnyy, N.M.</u> , <u>A.Z. Veksler</u> , and <u>A.I. Bulanova</u> (Sverdlovsk Branch of VNIIM). Method of Measuring Hysteresis Losses and Eddy Currents in Double Magnetization	104
Card 20/27	

86876

24,2200(1134,1158,1160)

S/105/61/000/001/003/007  
B012/B059

AUTHORS: Rudnyy, N. M., Veksler, A. Z., and Bulanova, A. I.  
TITLE: Measurement of the Losses in Ferromagnetic Materials  
Simultaneously Magnetized by Fields of Various Frequencies  
PERIODICAL: Elektrichestvo, 1961, No. 1, pp. 48-51

TEXT: In the present paper the method of loss measuring which was worked out by the authors is given for the most general case of a combined magnetization where the frequencies of the various field components are not multiple and not zero. It is shown that the method chosen in the case of combined magnetization for loss measurement should guarantee the measurement of the mean power, whereas the measuring instrument should be sufficiently inert not to respond to fluctuations of the measured quantity. The conditions on which losses can be measured may be given in various ways. The most expedient ones are: 1) frequencies  $f_1, f_2$  etc. and the amplitudes  $B_{m1}, B_{m2}$  etc. of the respective components of magnetic induction are given;

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Measurement of the Losses in Ferromagnetic  
Materials Simultaneously Magnetized by  
Fields of Various Frequencies

S/105/61/000/001/003/007  
E012/B059

2)  $f_1$  and  $f_2$  (or  $f_1$  and  $f_2-f_1$ ), highest and mean field strength amplitude, and mean value of the induction amplitude are given. The first way is more universal, the second one, however, the most agreeable in the case of magnetization by means of a modulated current. The device for loss measurement in the case of combined magnetization is based on the method of watt-meter operation. Fig. 2 illustrates the basic layout of this device. The low-frequency voltage component (up to 200 cps) can be measured by means of this instrument. A phase-sensitive voltmeter with two valves (Fig. 3) is used for measuring the voltage components of higher frequency. The device described here was used for measuring the losses in the cases of combined and of ordinary magnetization. It was found that the errors in loss measuring in the case of combined magnetization are greater than the errors in loss measurement by means of the watt-meter method in the case of raised frequencies and ordinary magnetization (Ref. 3). They amount to  $\pm 5\%$ . They are due to errors in the measurement of the secondary voltage by means of the phase-sensitive voltmeter.

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86876

Measurement of the Losses in Ferromagnetic  
Materials Simultaneously Magnetized by  
Fields of Various Frequencies

S/105/61/000/001/003/007  
B012/B059

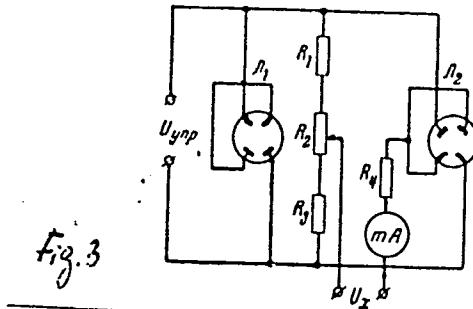
There are 4 figures and 3 references: 2 Soviet.

ASSOCIATION: Sverdlovskiy filial nauchno-issledovatel'skogo instituta  
metrologii im. Mendeleyeva (Sverdlovsk branch of the  
Scientific Research Institute of Metrology imeni Mendeleyev)

SUBMITTED: February 2, 1960

Card 3/5

86876

S/105/61/000/001/003/007  
B012/B059

Legend to Fig. 2: Basic diagram of the device for loss measuring with simultaneous magnetization by means of fields of various frequencies.  
1) Sound generator, 2) sound generator, 3) amplifier, 4) phase shifter,  
5) phase shifter, 6) phase-sensitive voltmeter, 7) voltmeter, 8) watt-  
meter, 9) amplifier, 10) wattmeter, 11) voltmeter, 12) investigated

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B012/B059

sample, 13) lever switch.

Legend to Fig. 3: Connection of the phase-sensitive voltmeter for 10 volts.  
 $R_1 = 6$  kilohms,  $R_2 = 0.5$  kilohms,  $R_3 = 6$  kilohms,  $R_4 = 1210$  ohms,  
1) control voltage.

✓

Card 5/5

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

VEKSLER, A.Z.; BULANOVA, A.I.; FALALEYNOVA, T.N.

Effect of inhomogeneous magnetization. Nov,nauch.-issl.rab.po.met.  
VNIM no.5:17-19 '64.  
(MIRA 18.3)

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

BULANOVA, N. I.; VENKUZAR, A.S.

Dependence of the magnetic path length on the field intensity  
(experimental study). Nov. nauch.-issl. rab. polimetr. VNIIIM  
no. 521-25 '64. (MINA 18:3)

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

RAFANOVA, R.Ya.; BULANOVA, A.V.

Studying chemical composition of an essential oil prepared from  
fresh and dried patchouli. Trudy VNIISNDV no.2:76-77 :54.

(Essences and essential oils) (Patchouli) (MLRA 10:7)

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

~~BUCHANAN, A. V.~~

~~RAFANOVA, R.Ya.; BULANOVA, A.V.~~

Patchouli oil obtained from distillation water. Trudy VNIISNDV  
no.2:77-78 :54.  
(MLRA 10:7)  
(Essences and essential oils) (Patchouli)

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

RAFANOVA, R.Ya.; BULANOVA, A.V.

Artificial drying of patchouli. Trudy VNIISNDV no.2:156-158 '54.  
(Patchouli) (MLRA 10:7)

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

RAFANOVA, R.Ya.; BULANOVA, A.V.

Clonal lavender oil. Trudy VNIISMDV no.4:93-98 '58.  
(Lavender oil) (MIRA 12:5)

RAFANOVA, R.Ya.; BULANOVA, A.V.

Chromatographic method for recovering linalyl acetate from  
essential oils of muscatel sage and lavender. Trudy VNIISNDV  
no.4:98-100 '58. (MIRA 12:5)

(Essences and essential oils)  
(Linaloöl)

BULANOVA, A.V.; RAFANOVA, R.Ya.

Composition and chemical transformations of Russian patchouli  
oil. Trudy VNIISNDV no.4:100-104 '58. (MIRA 1c:5)  
(Essences and essential oils)

GUSEVA, K.A.; RAFANOVA, R.Ya., kand.khim.nauk; BULANOVA, A.V.;  
VIREZUB, S.I.

Isolating sclareol and obtaining products from it having the  
odor of amber. Masl.-zhir.prom. 25 no.3:29-30 '59.

(MIRA 12:4)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskikh  
i natural'nykh dushistykh veshchestv.  
(Sclareol) (Perfumes, Synthetic)

BAG, A.A.; BLIZNYAK, N.V.; BULANOVA, A.V.; KUSTOVA, S.D.; CHERKAYEV, V.G.

Odorous substances from sclareol. Report No.2: Possibility for  
converting the lactone 1,1, 6, 10-tetramethyl-6-oxy-5-methylene-  
carboxydecalin into 1, 1, 6, 10-tetramethyl-6-oxy-5-( $\beta$ -oxy)-  
ethyldecalin by catalytic hydrogenation. Trudy VNIISNDV no.5:  
14-16 '61.

(Odorous substances) (Naphthalene) (MIRA 14:10)

BULANOVA, F.G.

Fauna and ecology of gadflies (Diptera, Tabanidae) in the Udmurt  
A.S.S.R. Med.paraz.i paraz.bol. no.1:36-38 '62. (MIRA 15:5)

1. Iz kafedry biologii Izhevskogo meditsinskogo instituta.  
(UDMURT A.S.S.R.—HORSEFLIES)

BULANOVA, F. G.

Horseflies in the Udmurt A.S.S.R. Zool. zhur. 42 no.4:627-628  
'63. (MIRA 16:7)

1. Department of Biology, Izhevsk Medical High School.  
(Udmurt A.S.S.R.--Horseflies)

BULANOVA, G.V.

Pseudotuberculosis in rodents in a large city. Tez.i dokl.konf.Irk.  
gos.nauch.-issl.protivochum.inst. no.1:9-10 '55. (MIRA 11:3)  
(RODENTIA--DISEASES AND PESTS) (LUNGS--DISEASES)

BULANOVA, G.V.

Some epidemiological and clinical peculiarities of plague in  
its natural marmot foci in Bayan Khongor District of the Mon-  
golian People's Republic. Izv. Irk.gos.nauch.-issl.protivochum.  
inst. 20:77-85 '59. (MIRA 13:7)  
(BAYAN KHONGOR DISTRICT (MONGOLIA)--PLAQUE)

NEVEL'SON, M.I.; NIKITIN, A.I.; YANISHEVSKIY, V.V.; BOYKO, G.G.; KUZNETSOV,  
N.I.; ~~BULANOVA, I.A.~~; GORSHKOV, V.I.; KATSMAN, I.A.; KUKAYEVA, YE.V.;  
RYZHOOVA, V.V.; TURBOVA, V.I.; CHEREDEYEVA, Ye.M.; KOSHEL'KIN, M.V.

Development of highly efficient ventilator models ORGRES operating  
according to a 0.68-161° system for electric power plants. Prem.  
energ. 18 no.7:8-9 Jl '63.

(MIRI 16:9)

(Electric power plants--Electric equipment)  
(Fans, Electric)

L 16916-65 EWT(m) DIAAP

ACCESSION NR: AP4047846

S/0186/64/008/005/0621/0623

AUTHOR: Bulanova, I. D.; Vorob'yev, A. M.

TITLE: Extraction of protactinium from hydrochloric acid solutions by tributylphosphate

SOURCE: Radiokhimiya, v. 6, no. 5, 1964, 621-623

TOPIC TAGS: protactinium extraction, tributylphosphate, protactinium solvation

ABSTRACT: The authors call attention to the fact that, together with the higher alcohols and certain ketones, tributylphosphate (TBP) is an effective solvent for protactinium. The purpose of the present article was to determine, experimentally, the composition of a protactinium complex extracted from hydrochloric acid solutions by means of TBP. The experimental technique employed is not described in this paper. The results of the study are presented in tabular and graph form. The authors found that at  $[H^+]$  and  $[Cl^-]$  concentrations from 2 to 6 M and  $\mu = 5-6$ , protactinium was extracted into the organic phase in the form of  $H_3PaOCl_4$ , solvated by 2 molecules of TBP. Orig. art. has: 3 tables and 4 figures.

Card 1/2

L 16916-65

ACCESSION NR: AP4047846

ASSOCIATION: None

SUBMITTED: 01Feb64

ENCL: 00

SUB CODE: IC, GC

NO REF SOV: 002

OTHER: 003

Card 2/2

ACCESSION NR: AP4047847

S/0188/64/008/005/0823/0828

B

AUTHOR: Bulanova, I. D.; Vorob'yev, A. M.

TITLE: Extraction of protactinium from hydrochloric acid solutions by methylisobutyl ketone.

SOURCE: Radiokhimiya, v. 6, no. 5, 1964, 623-626

TOPIC TAGS: protactinium extraction, methylisobutyl ketone, thorium decay

ABSTRACT: While it is known that the most effective extracting agents for protactinium are the higher alcohols and certain ketones, a number of problems remain to be solved in this area, there being, in particular, no consensus with respect to the composition of the extracted complexes. There are, moreover, no available data in the technical literature regarding the solvation of protactinium compounds by methylisobutyl ketone. The authors therefore studied protactinium extraction from hydrochloric acid solutions by methylisobutyl ketone and investigated the extracted complexes. The protactinium-233 was derived from 2 g of metallic thorium irradiated for 2 days in a reactor with a stream of thermal neutrons ( $10^{11}$  neutrons/cm<sup>2</sup> sec); the irradiated thorium was then dissolved in hydrochloric acid, with the protactinium-233 being isolated by extraction methods. The isotope

Card 1/3

ACCESSION NR: AP4047847

obtained by the techniques described in the paper was sufficient for 5 months' work, with the protactinium subjected to supplementary purification as it decayed. The initial concentration of Pa in the aqueous phase was 10-12 M, with equilibrium achieved after 2-5 minutes by open-tube shaking at room temperature. Phase separation after equilibrium was accelerated by centrifugation. The method for determining the isotope content is described in detail in the article. The Pa distribution factor was calculated as the ratio of the measured protactinium concentrations (activity per unit volume in the organic and aqueous phases). A full discussion of the results of the experiments conducted by the authors is given in the article; they may be summarized as follows: 1. The data obtained confirm the effectiveness of methylisobutyl ketone as a solvent for the quantitative extraction of Pa 233 from hydrochloric acid solutions. 2. With  $[H^+]$  and  $[Cl^-]$  concentrations of from 2 to 6 M and  $\mu = 5-6$ , protactinium is extracted into the organic phase in the form of  $H_2PaOCl_5$ . Orig. art. has: 3 tables and 4 figures.

Cold 2/3

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

ACCESSION NR: AP4047847

ASSOCIATION: None

SUBMITTED: 01Feb84

NO REF SOV: 001

ENCL: 00

SUB CODE: IC, GC

OTHER: 004

Card 3/3

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

MASLENNIKOV, K.N., nauchnyy sotrudnik; ZAYTSEVA, Ye.V., nauchnyy sotrudnik;  
KANTER, D.TS., nauchnyy sotrudnik; OBUKHOVA, R.N., nauchnyy sotrud-  
nik; BULANOVA, I.G., nauchnyy sotrudnik; GORDEYEV, N.A.; SURNINA,  
N.M.

"Xylital 0-15" preparation for the avivage of viscose staple fi-  
bers produced by the cotton spinning method. Tekst.prom. 24 no.1:  
40-43 Ja '64. (MIRA 17:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo  
volokna (for Maslennikov, Zaytseva, Kanter, Obukhova, Bulanova).
2. Glavnyy inzh. Yakhromskoy pryadil'no-tkatskoy fabriki (for Gor-  
deyev).
3. Zaveduyushchiy proizvodstvennoy laboratoriyye Yakhrom-  
skoy pryadil'no-tkatskoy fabriki (for Surnina).

BULANOVA, I.V.

KASHINTSEVA, N.S.; GIL'GUT, Ye.A.; BULANOVA, I.V.

Study of tetanus toxins and anatoxins grown on casein media. Zhur. mikrobiol. epid. i immun. 28 no.4:10-14 Ap '57. (MLR 10:10)

1. Iz Institute epidemiologii i mikrobiologii imeni N.F.Gamalei AMN SSSR.

(TETANUS

anatoxins & toxins grown on casein medium. qualities)

USSR / Microbiology. Anaerobic Bacilli.

F-6

Abs Jour: Ref Zhur-Biol., No 15, 1958, 72199.

Author : Kashintseva, N. S.; Gil'gut, Ye. A.; Bulanova,  
I. V.

Inst : Not given.

Title : Concentrated Purified Tetanus Anatoxin and Its  
Immunological Properties.

Orig Pub: Zh. mikrobiol., epidemiol. i immunobiologii, 1957,  
No 10, 89-94.

Abstract: No abstract.

Card 1/1

BULANOVA, K. N.; Cassovsky, L. N.; Shvarts, Z. N.

The State Optical Inst (G.O.A.) Leningrad, Min Defense USSR

"Dependence of light dams of the eye on the size of colour sources of light"

SOURCE: Doklady Akademii Nauk, Vol 53, No 6, 1947

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

BULANOVA, K. R.

Spectral sensitivity of the central part of the retina. K. R.  
Bulanova [C. R. Acad. Sci. U.R.S.S., 1953, 91, 1333-1336].  
Absolute thresholds for a 1' test field in the dark-adapted eye  
were determined at 435, 540, 579, and 700 mμ, and locations  
7°, 15°, 22°, 37°, 1° 15°, 2° 30°, 5°, and 10° from the foveal centre.  
G. S. BRINDELEY

R<sup>W</sup>  
10-12-51

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

BULANOVА, K. N.

USSR/Physics - Light Threshold, Eye 1 Nov 53

"The Light Thresholds in the Foveal Portion of the Retina," K. N. Bulanova

DAN SSSR, Vol 93, No 1, pp 29-30

Purpose is to det the nature of the light thresholds (i.e. the light-perception at threshold) and to measure them quantitatively in the fovea. Notes that the existence or non-existence of an achromatic threshold in the fovea is still in doubt (N. I. Pinegin, Probl Fiziol Optiki (Problems of Physiological Optics), No 8 (1953); K. N. Bulanova,

275T91

ibid.). Cites related study of L. N. Gassovskiy, K. N. Bulanova, Z. M. Shvarts (DAN, 58, No 6, 1947). Presented by Acad A. N. Terenin 3 Sep 53.

USSR/Medicine - Biophysics

Card 1/1      : Pub. 22 - 9/41  
Authors      : Bulanova, K. N., and Luizov, A. V.  
Title        : Threshold duration of blacking-out a point source of light  
Periodical    : Dok. AN SSSR 98/2, 205-206, Sep 11, 1954  
Abstract     : Experiments with the threshold duration of the blacking-out of a periodical point-light source with respect to its brightness and the duration of sparks are described. Two references (1949 and 1953).  
Institution : ...  
Presented by : Academician A. N. Terenin, April 12, 1954

Translation D-177062, 29 Nov 54

BULANOVA, K.N.; LUIZOV, A.V.

Influence of certain factors on the visibility of flashlights in navigation. Dokl. AN SSSR 109 no.3:511-514 Jl '56. (MIRA 9:10)

1. Predstavлено академиком А.Н. Терениным.  
(OPTICS, PHYSIOLOGICAL)

612.843.613

✓ 2897. EFFECT OF SOME FACTORS ON THE VISIBILITY OF  
FLASH LIGHTS. K.N.Bulanova and A.V.Luzov.

Dokl.Akad.Nauk SSSR, Vol. 109, No. 3, 511-14 (1956). In Russian.

It was found that the length of period T of short periodic flashes has no effect on the relationship between the threshold brightness of the flash and its duration, at least within the T range 1-5 sec. For all background brightness values investigated the dependence of the threshold brightness quantity y and the duration of the flash is  $y = aT + b$  ( $y = E_T$ , i.e. the product of brightness and the duration at the visibility threshold). The parameters a and b depend almost exclusively on background brightness (and very little on the colour).

P. Lachman

2

Red

LUIZOV, A.V.; BULANOVA, K.N.

Effective brilliance depending on background brightness. Svetotekhnika  
3 no.7:17-19 J1 '57.  
(MERA 10:8)

1.Gosudarstvennyy opticheskiy institut.  
(Photometry)

LUIZOV, A.V., kand.fiz.-mat. nauk; BULANOVA, K.N., inzh.

Threshold brilliance depending on dimensions of the source and flash duration. Svetotekhnika 4 no.10:17-20 0 '58. (MIRA 11:10)

1.Gosudarstvennyy opticheskiy institut.  
(Light)

LUIZOV, A.V., kand.fiz.-mat.nauk; BULANOVA, K.N., inzh.

Threshold difference in the brillance of consecutive flashes.  
Svetotekhnika 5 no.7:22-25 J1 '59. (MIRA 12:9)  
(Lightships)

MIKHALEVSKAYA, Ye.S.; VOLKOV, O.S.; BULANOVA, L.P.; BERKOVICH, T.M.

Effect of the water-cement factor on the kinetics of cement and  
asbestos cement hydration. Trudy NIIAsbestsementa no.15:31-37  
'62. (MIRA 16:7)  
(Cement) (Asbestos cement)

BERKOVICH, T.M.; ISAYEVA, O.A.; BULANOVA, L.P.; LYAPINA, R.V.

Capillary water saturation of asbestos cement and its effect  
on the reinforcing properties of chrysotile-asbestos fibers.  
Trudy NIIAsbesttsementa no.19:3-20 '65.

(MIRA 18:9)

BURCHAK, G.P., dots.; BULANOVA, N.F., assistant; ZYLEV, B.V.,  
dots.; PIAVDIN, Zh.L., dots.; KUROVA, A.V., red.

[Methods manual or the solution of problems in theoretical mechanics; dynamics] Metodicheskoe posobie po resheniju zadach teoreticheskoi mekhaniki; dinamika. Moskva, Mosk. in-t inzhenerov zhel-dor. transp., 1962. 163 p. (MIRA 18:8)

BULANOV<sup>A</sup>, N.K.

BULANOVA, N.K.; KERTSMAN, L.I.; PLISETSKAYA, M.A.; SOKHOR, N.M.

Medical and sanitary services for industrial workers of Leningrad  
District in Moscow. Zdrav.Ros.Feder. 1 no.6:11-15 Je '57.

(MIRA 10:8)

1. Iz sanitarno-epidemiologicheskoy stantsii Leningradskogo rayona  
Moskvy  
(MOSCOW--INDUSTRIAL HYGIENE)

BULANOVA, N.K.; PLISETSKAYA, M.A. (Moskva)

Improvement of working conditions on the vibrating conveyer  
section. Gig.truda i prof.zab. 3 no.4:42-44 Jl-Ag '59.  
(MIRA 12:11)

1. Sanitarno-epidemiologicheskaya stantsiya Leningradskogo  
rayona.

(CLOCKMAKING AND WATCHMAKING--HYGIENIC ASPECTS)

BULANOVA, O.N.

Ca

PROCESSED AND PROPRIETARY DATA

The effect of blood transfusion on the blood gases in traumatic shock. G. V. Derviz, O. N. Bulanova, and A. G. Stepanenko. *Klin. Med. (U.S.S.R.)* 21, No. 9, 51-60 (1943). - Expts. on dogs are reported. After traumatic shock an oxygen deficiency is produced in the venous blood. This causes a shift in the acid-base system, with the appearance of acid substances in the blood, and a lowering of the alk. reserve. The transfusion of massive doses of blood gradually restores a normal balance of the gases and a normal condition of the acid-base system.  
S. Gottheb

1193

ASA SIA METALLURGICAL LITERATURE CLASSIFICATION

SECOND LEVEL

SECOND HIER ONLY ONE

ASSISTANT

SECOND LEVEL

SECOND HIER ONLY ONE

ASSISTANT

BULANOVA, O.N.

BULANOVA, O.N. (Moscow); DEROVIZ, G.V., professor, zaveduyushchiy.

Activity of catalase in the blood in leukemias. Klin.med. 31 no.8:46-51  
Ag '53. (MLRA 6:11)

1. Biokhimicheskaya laboratoriya TSentral'nogo ordena Lenina instituta gema-tologii i perelivaniya krovi.  
(Leukemia) (Oxidases) (Blood)

Organic acids of blood of animals in the stage of dying from loss of blood and in the consequent restoration of life functions of the organism. O. N. Bulanova (Acad. Med. Sci. U.S.S.R., Moscow). *Biofizika* 1954, 39, 590-8 (1954).—Dogs were subjected to pentobarbital-ether narcosis and exsanguinated from the femoral artery. Following a given period of "crit. death" they were brought to life by the method of Negovskii. Arterial blood samples were taken under normal conditions, under narcosis, and at various stages of crit. dying and life restoration. Analyses were made at once and also 20 min., 1, 1½, 3, 24, and 48 hrs. later. The org. acids of plasma were detd. electrometrically; lactic acid was detd. by the method of Friedman, *et al.* Increase in the amt. of org. acid becomes evident when the loss of blood is approx. 50%. The increased org.-acid content of the blood plasma persists during the period of clinical death and at the initiation of revivification, regardless of the restoration of the blood circulation and of the application of artificial respiration. The content of org. acids in the blood plasma begins to recede only at the time natural respiration sets in, reaching the normal level about 3 hrs after life restoration. In the majority of expts. the quantity of org. acids in the blood is detd. by the duration of deep hypoxia. Under normal conditions lactic acid constitutes 20% of the org. acids of the blood. During the state of agony it reaches 37%. The quantity of X-acids in the blood plasma markedly increases as the state of hypoxia grows in intensity and exceeds the normal level by 85% at the time independent respiration is restored. B. S. Leyne

"APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4

BULANOVA, O.N.

Twentieth anniversary of the Laboratory of Experimental Physiology  
of Resuscitation of the Academy of Medical Sciences of the U.S.S.R.  
Vest.AMN SSSR 12 no.2:79-83 '57. (MIKA 10:10)  
(RESUSCITATION)

APPROVED FOR RELEASE: 06/09/2000

CIA-RDP86-00513R000307420010-4"

BULANOVA, O.N.; KISELEVA, K.S. (Moskva)

Effect of sodium bicarbonate on the restoration of vital functions after clinical death caused by blood loss. Pat. fiziol. i eksp. terap. 3 no.2:59-67 Mr-Ap '59. (MIRA 12:6)

1. Iz laboratorii eksperimental'noy fiziologii po ozhivleniyu organizma (zav. - prof.V.A.Negovskiy) AMN SSSR.  
(CARBONATES, eff.

sodium bicarbonate, on restoration of vital funct. after clin. death induced by desanguination in dogs (Rus))  
(RESUSCITATION

eff. of sodium bicarbonate after clin. death induced by desanguination in dogs (Rus))

BULANOVA, O.N.; ZAKS, I.O.

(Moskva)

Hypoxia and acid-base equilibrium in prolonged artificial circulation produced by direct heart massage. Pat. fiziol. i eksp. terap. 6 no.6:17-22 N-D'62 (MIRA 17:3)

1. Iz laboratorii eksperimental'noy fiziologii po ozhivleniyu organizma ( zav. - prof. V.A. Negovskiy) AMN SSSR.

BULANOVA, O.N.; ZAKS, I.O. (Moskva)

Blood substitution and oxidation of the intermediate products  
of metabolism in resuscitation after clinical death. Pat.  
fiziol. i eksp. terap. 7 no.4:40-45 Jl-Ag '63.

1. Iz laboratorii eksperimental'noy fiziologii po ozhivleniyu  
organizma (zav.- prof. V.A. Negovskiy) AMN SSSR. (MIRA 17:9)

BULANOVA, O.N.

Hypoxia and acid-base equilibrium in dying from loss of blood under conditions of deep hypothermia and in the subsequent restoration of vital functions following a two-hour clinical death. Pat. fiziolog. i eksp. terap. 9 no.2:26-30 Mr-Ap '65. (MIRA 18:5)

1. Laboratoriya eksperimental'noy fiziologii po ozhivleniyu organizma (zav. - prof. V.A.Negovskiy) AMN SSSR, Moskva.

BULANOVA, O.N.; ZAKS, I.O.

Acid-base equilibrium in the organism in dying from hemorrhage  
and in subsequent restoration of life functions. Pat. fiziol.  
i eksp. terap. 9 no.5:84-86 S-0 '65. (MIRA 19:1)

1. Laboratoriya eksperimental'noy fiziologii po ozhivleniyu  
organizma (zav. - prof. V.A. Negovskiy) AMN SSSR, Moskva.  
Submitted September 18, 1963.

BULANOVA, S.I.; SMILEVICH, V.B.; SHERMAN, A.Sh.

Role of a dispensary for tuberculosis control in the detection  
of lung cancer. Vop. onk. 11 no.3:85-89 '65.

(MIRA 18:6)

1. Iz protivotuberkuleznogo dispansera No.11 Moskvy (glavnnyy  
vrach - kand. med. nauk A.Sh. Sherman) i 1-go khirurgicheskogo  
otdeleniya (zav. - doktor med. nauk B.Ye. Peterson) Instituta  
eksperimental'noy i klinicheskoy onkologii AMN SSSR (dir. -  
deystvitel'nyy chlen AMN SSSR prof. N.N. Blokhin).

5 (4)

AUTHORS: Kazanskiy, B. A., Landsberg, G.S. (Deceased), SOV/62-59-9-15/40  
Aleksanyan, V. T., Bulanova, T. A.,  
Liberman, A. L., Mikhaylova, Ye. A., Plate, A. F., Sterin, Kh.Ye.,  
Ukholin, S. A.

TITLE: Investigation of the Composition of the Fraction With a Boiling Point Between 150 and 250° of the Emba Crude Petroleum

PERIODICAL: Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk, 1959, Nr 9, pp 1612 - 1622 (USSR)

ABSTRACT: An attempt is being made to apply the combined investigation method for benzines (Ref 1) to the investigation of the petroleum fraction with a boiling point between 150 and 250° of the Emba crude petroleum. The petroleum investigated came from the Koschagylskoye deposit. It was proved that this fraction contains 12.6% of aromatic and 13.0% of hexahydroaromatic hydrocarbons. In the aromatic fraction 29 different hydrocarbons were identified. The quantitative division in groups of the aromatic hydrocarbons boiling in this range was carried out with characterization of the arrangement of the side-chains on the benzene ring or the corresponding cyclohexane ring and that for the multi-cyclic according to the arrangement of the rings. By this method

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Investigation of the Composition of the Fraction With SOV/62-59-9-15/40  
a Boiling Point Between 150 and 250° of the Emba Crude  
Petroleum

the authors succeeded in establishing the composition of the aromatic compounds up to 70% and that of the hydroaromatic compounds up to 46%. In the paraffin-naphthene part of the fraction the presence of naphthene with two different substituents in the same carbon atom of the cyclohexane could be established (mixed substitution). The limiting into narrower fractions was possible at the paraffin-naphthenes by investigating the specific gravities, the refractive index and the aniline point of these fractions. In figures 1 and 2 the paraffin-naphthene fractions are identified and tables 1-6 contain the results of the analysis. Table 7 gives the results of the distillation of the paraffin-cyclopentane fraction of the Ligroin applying the coefficient proposed by P. S. Maslov (Ref 11). There are 2 figures, 7 tables, and 11 references, 10 of which are Soviet.

Card 2/3

Investigation of the Composition of the Fraction With  
a Boiling Point Between 150 and 250° of the Emba Crude SOV/62-59-9-15/40  
Petroleum

ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo Akademii  
nauk SSSR (Institute of Organic Chemistry imeni N. D. Zelinskogo  
of the Academy of Sciences, USSR). Komissiya po spektroskopii  
Akademii nauk SSSR (Committee of Spectroscopy of the Academy  
of Sciences, USSR)

SUBMITTED: January 4, 1958

Card 3/3

*con*

PROBLEMS AND PROSPECTS 1921

Thermal decomposition of lead formate and formic acid on metal lead surface. L. K. Freudlin and T. F. Bulanov. Bull. Acad. sci. U. R. S. S., Class sci. math. nat. Nauk. chesk. 1937, 555 (7 am English 1938). The products as well as the kinetics of the decompr. of Pb formate and HCOOH on Pb were studied. The theory of intermediate salt formation is supported by the similarity of products obtained from the decompr. of HCOOH on Pb and Pb formate on Pb. The catalytic effect of Pb on the decompr. of HCOOH is discussed. The absence of volatile liquid products during the decompr. of HCOOH on surfaces of metallic catalysts is explained by the action of the final products upon the intermediate salt. The results were treated according to Arrhenius' equation for first-order reactions giving the energy of activation. Below 235° the gaseous products of  $(\text{HCOO})_2\text{Pb}$  decompr. changed in compn. About 284°, the gaseous compn. was const., and above this temp. range the evolution was so rapid as to invalidate observations. The av. value for the energy of activation from three curves is 20,000 cal./mol.  $(\text{HCOO})_2\text{Pb} \sim \text{PbO} + \text{CO}_2 + \text{CH}_3\text{O}$  predominates at 240-260° at first, later  $(\text{HCOO})_2\text{Pb} \sim \text{Pb} + 2\text{CO}_2 + \text{H}_2$

leads at all temps. because of the accumulation of Pb. This is shown by the decreasing value of  $\text{CO}_2/\text{H}_2$  with 2 as the limit at 240°. Mammich and Gelmann found that  $2\text{CH}_3\text{O} \rightarrow \text{HCOOC}_2\text{H}$ , which scheme is accepted. The catalytic decompr. of HCOOH was first studied by Sabatier and Marthe and later by Adkens and Nissen, who established the correlation between the method of prepn. of  $\text{Al}_2\text{O}_3$  and  $\text{HCOOH} \rightarrow \text{CO}_2 + \text{H}_2$  or  $\text{HCOOH} \rightarrow \text{CO} + \text{H}_2\text{O}$ . Adadurov studied the decompr. of HCOOH on charcoal impregnated with PbO. The authors agree that exclusively dehydrogenation results with charcoal or PbO catalyst taken separately. 190 ppm. on charcoal catalyzes dehydrogenation mainly, and at 235-325° there is a fourfold increase; 327° marks the peak of this tendency, while the m.p. of Pb is 325°. Solid and molten Pb cause  $\text{HCOOH} \rightarrow \text{CO}_2 + \text{H}_2$ , for which data obtained by passing HCOOH over powd. Pb yield 32,000 cal./mol. as the energy of activation. A film of Pb formate upon Pb is deemed significant as a catalyst. —Sergius Kobernick

ESTERIFICATION OF ALCOHOLS BY CATALYTIC DEHYDROGENATION

D. N. Vaskevich and T. E. Bulanova. *J. Gen. Chem. (U.S.S.R.)* A, 1091-7 (in English, 1007) (1968). - MeOH, EtOH, iso-BuOH and iso-AmOH were dehydrogenated at atm. pressure and 270-30° by passing mixt. of 2 alcohols in all the possible combinations at the rate of 12 to ml./hr. over the catalyst Cu-Mn (3.5:1) activated with Ag. The latter was prep'd. by treating the nitrates with aq. NaOH, drying the washed ppt. at 105° and activating in EtOH at 300°. After the dehydrogenation for 10 hrs., the catalyst showed no signs of deactivation. The preliminary results in the analysis of the catalysts indicate the presence of 30% esters, considerable unaltered alcohols, some aldehydes and practically no acids. The general trend of the reaction is toward the formation of fewer esterification products than is theoretically possible and the acidic radical contains the smaller no. of C atoms. Thus, a mixt. of MeOH and iso-AmOH gave chiefly iso-Am formate and some Me isobutyrate instead of the 4 theoretically possible esters, while iso-BuOH with iso-AmOH formed practically only iso-Am isobutyrate. MeOH with iso-BuOH gave iso-Bu formate and a little

<sup>1</sup> Me isobutyrate. The investigation is being continued.  
Twenty references. Chas. Blanc

Reaction of sodium amide with aromatic ketones in the molten state. L. Kh. Freidlin and T. F. Bulanova, *J. Gen. Chem. (U.S.S.R.)*, 9, 299-303 (1939).—Analogous to the reaction between  $\text{NaNH}_2$  (I) and salts of organic acids the aromatic ketones with I in the molten state undergo cleavage of the CO group to give  $\text{NaHCN}$  (II), the corresponding hydrocarbons and probably  $\text{NaOH}$ , according to the equation:  $\text{RCOR}' + 2\text{NaNH}_2 \rightarrow \text{NaHCN} + \text{NaOH} + \text{RH} + \text{R}'\text{H}$ . In addn., considerable C as well as small量 of HCN,  $\text{NaCN}$ , dicyanodiimide and  $\text{NH}_3$  are formed, probably due to secondary processes. The reaction, strongly exothermic, proceeds vigorously in all cases at temps. below 150° and for the diketones is often very violent. Benzenophenone, benzil and benzoin with I at 95-110° give  $\text{C}_6\text{H}_6$ , II and  $\text{NH}_3$ . Fluorone and phenanthrenequinone give  $\text{Ph}_2$  in addn. to other products. Anthraquinone with I gives a brown, amorphous product, not investigated. John Livak

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## Lab. Organic Catalysis, Inst. Org. Chem., AS USSR

**APPROVED FOR RELEASE: 06/09/2000**

CIA-RDP86-00513R000307420010-4"

Bulanova, T. F.

Bazhulin, P. A., Sterin, Kh. E., Bulanova, T. F., Sоловьева, О. П. CA: 42-6238/1  
Turova-Pollak, M. B. and Kazanskiy, B. A.  
(P. N. Lebedev Phys. Inst. and Inst. Org. Chem., Acad. Sci. USSR, Moscow and  
Moscow State Univ.)  
Izvest. Akad. Nauk SSR Otdel. Khim. Nauk 1946, No. 1, 7-18  
Optical investigation of hydrocarbons. IV. Raman spectra of cycloparaffins.

Catalytic hydrogenation of cyclopentane hydrocarbons with ring opening. VI. Hydrogenation of cyclopentane in the presence of platinized carbon. B. A. Kuzanski and T. F. Il'ianova. Inst. Org. Chem. Acad. Sci. U.S.S.R., Moscow. *Bull. Acad. Sci. U.R.S.S. Clas. sci. chim.* 1947, 29-40 (in Russian); cf. *C.A.* 29, 7950; 33, 9263; 37, 3035. — (1) The catalyst (40 ml.), having an activity characterized by complete hydrogenation of  $\text{C}_5\text{H}_8$  at a space velocity  $v$  of 0.3 and 80°, dehydrogenation of cyclohexane at 300° and the same  $v$ , was active in the hydrogenation of cyclopentane (I) only when virgin; under these conditions, at 225° and  $v = 0.1$ , the initial activity corresponded to 44.6% hydrogenation of I (equivalent to 1.18 g.  $\text{C}_5\text{H}_8/\text{hr.}$ ) but fell in subsequent runs, reaching a const. activity of about 0.1 g.  $\text{C}_5\text{H}_8/\text{hr.}$  after about 9 hrs. Catalysts previously tested with  $\text{C}_5\text{H}_8$  or with cyclohexane were inactive in the hydrogenation of I at 200° and  $v = 0.1$ . (2) The catalyzate obtained at 200° (in several runs of 1.5-2 hrs. each) was fractionated into  $b_{100}^{\circ}$  36.9, 36.49, 49°, and residue, in the ratio 70.5, 0.3, 10.6, and 3.6%; the 1st and the 3rd fractions were shown to consist of pure  $\text{C}_5\text{H}_8$  and I, resp. The gas was  $\text{H}_2$ , 97.48% satd. hydrocarbons 3-2%; hence, cracking products are practically absent. The rates of formation of  $\text{C}_5\text{H}_8$  at a feeding rate of 2.24 g. I/hr., outgoing gas equiv. to 11./hr., at 225, 237, 250, 262, and 275°, were 0.11, 0.34, 0.68, 1.37 and 2.00 g./hr., resp.; at 202°, the yield of  $\text{C}_5\text{H}_8$  was practically independent of the rate of feeding (2.24-7.00 g./hr.); at 275°, with I fed at 2.24, 5.01, 8.69, 0.16 g./hr., the yields of  $\text{C}_5\text{H}_8$  were 2.00, 3.27, 3.04, 3.05, 3.11 g./hr., i.e., a rate of 2 g./hr. is insufficient for full utilization of all the active centers of the catalyst; at 200°, 8.5-0.6 g. I/hr., the yield of  $\text{C}_5\text{H}_8$  was about 7 g./hr. The apparent activation energy of the reaction is about 36 kcal./mole.

(3) Under the same conditions,  $\text{C}_5\text{H}_8$  suffers no change over the same catalyst. (4) For purposes of analysis of mixts. of I and  $\text{C}_5\text{H}_8$ ,  $\pi_0^{\circ}$ ,  $\Omega^{\circ}$ , on-line points, and sp. vol. were detd.; selected points: I 80.81, 73.40, 57.43, 40.28, 13.94%,  $\pi_0^{\circ}$  1.4015, 1.3934, 1.3853, 1.3778, 1.3654,  $\Omega^{\circ}$  0.7323, 0.7117, 0.6917, 0.6712, 0.6434; on-line point 21.7, 31.3, 40.1, 50.0, 63.5%. VII. Hydrogenation of methylcyclopentane in the presence of platinized carbon and of nickel on aluminum oxide. B. A. Karanski and Z. A. Rumyantseva. *Ibid.* 183 (8). (1) With 16 ml. of a platinized C catalyst (20% Pt, reduced with  $\text{H}_2$  up to 310°), having an activity characterized by 100% hydrogenation of  $\text{C}_5\text{H}_8$  at 150° and  $v = 0.22$  and by 43% hydrogenation of cyclopentane at 275°,  $v = 0.1$ , the catalyzate obtained in hydrogenation of methylcyclopentane (II) at 200°,  $v = 0.18$ , outgoing gas 2 l./hr., contained 40-41% paraffin hydrocarbons of which 2-methylpentane was 0.4%, 3-methylpentane 20%, and hexane 11%; this catalyst was fractionated into  $b_{100}^{\circ}$  59.0-9.6, 59.5-80.5, 80.5-2.8, 62.8-4.0, 64.0-8.8, 68.5-70.9, 70.9-1.5, 71.5-1.75, in the ratio 1.13, 17.60, 13.07, 3.37, 3.40, 12.00, 3.39, 35.87%; residue and losses 2.29%; fractions 1-4 were almost 100% paraffins, fractions 7 and 8 almost all II, fraction 5 contained about 70% paraffins, fraction 6, 15%. The catalyzate obtained at 300-3° was fractionated into  $b_{100}^{\circ}$  55.0-9.5, 59.5-80.5, 60.5-2.4, 62.4-3.0, 63.0-8.0, 68.0-9.1, in the ratio 4.70, 49.74, 10.70, 5.75, 10.30, 8.90%; residue and losses 3.85%; the 1st 3 fractions consist almost entirely of

paraffins, fractions 6 and 7 contain about 60% paraffins; thus, the total catalyze contains 93% paraffins of which 2-methylpentane is 68%, 3-methylpentane 20%, hexane 12%. The gas evolved at 300°-303° was II, 96.9, CH<sub>4</sub>, 12%, O<sub>2</sub> 04., N<sub>2</sub> 1.4%. The catalyst obtained at 317-320° consisted on the average of 75-77% paraffins but the paraffin content attained 100% in individual expts. The corresponding fractions of the 200, 300-3, and 317-20° catalysts were united and refractionated, giving 20° catalysts which were united and refractionated, giving b.p. 52.0-8.5° (8.32%), b.p. 50.5-0.8 (38.24), b.p. 61.0-2.75 (13.70), b.p. 62.5-4.0 (13.12), b.p. 65.0-8.8 (6.40), b.p. 68.8-71.4 (11.68), b.p. 71.4-1.7 (7.62), residue 0.90%; by the phys. consts., the paraffins consist of 2-methylpentane 67%, 3-methylpentane 23%, hexane 11%. The residue contained 2% C<sub>6</sub>H<sub>6</sub> (0.05% of the total catalyze). By the Raman spectra, the compn. is 72, 20, and 8%; and the contents of the sep. fractions 2, 3, 4, 5 + 6, and 7: 2-methylpentane, 100, 00, 23, 0, and 0%; 3-methylpentane, 0, 40, 77, 0, and 0%; hexane, 0, 0, 0, 30, and 10%; II, 0, 0, 0, 70, and 90%. (2) With a Ni catalyst on Al<sub>2</sub>O<sub>3</sub>, characterized by 100% hydrogenation of C<sub>6</sub>H<sub>6</sub> at 150° and  $\nu = 0.19$  and by 77% dehydrogenation of cyclohexane to C<sub>6</sub>H<sub>6</sub> at 290° and  $\nu = 0.19$ , II undergoes much cracking. At 240°, the gaseous products constitute 40% of the II introduced. At 260°,  $\nu = 0.16$ , the amt. of II cracked dropped from 46.8 to 33.0% in 4 consecutive runs, decreasing with progressing poisoning of the catalyst. Typical compn. of the gas: II, 67.6, paraffin hydrocarbons 32.6%. The collected catalyze,  $\nu^{\text{d}}_1$  1.3080 (increasing from 1.3070 to 1.4098 in 4 consecutive runs), d<sup>25</sup><sub>4</sub> 0.7080, ailine point 44.7°, was fractionated into b.p. 27.4-40.5° (34.74%), 40.5-72.0° (57.90), residue 3.60%; losses 3.00%; by the phys. consts., it contains 35% paraffins; the residue contains C<sub>6</sub>H<sub>6</sub>. At 280°,  $\nu = 0.15$ , cracking attained 60%,  $\frac{1}{2}$  of the catalyst, 1.4040. (3) With the same Ni catalyst, 3-methylpentane at 200°,  $\nu = 0.15$ , gave 42% catalyze, b.p. 24.0-62.8°,  $\nu^{\text{d}}_1$  1.3782, d<sup>25</sup><sub>4</sub> 0.6670, ailine point 67.1%; the gas was II, 47.0, sdd. hydrocarbons

dimethylpentane and I, with the latter predominating (50-60% of the fraction). Thus, the main reaction consists in ring opening at the 3,4- or the 4,5-bond. The gas consists of 95.0% II, and 4-4% sdd. hydrocarbons, i.e., cracking is insignificant. (2) 1,3-Dimethylcyclopentane (II), at 305°, vol. rate 0.27, gave a catalyze contg. 56.00% paraffins, gas 97.7% II, 2% C<sub>6</sub>H<sub>6</sub>, at 275°, vol. rate 0.2, the catalyze contained only 25% paraffins. The liquid obtained at 305° was fractionated under 750 mm. into b.p. 71.5-80.0° (16.1%), 80.0-81.0° (0.0%), 81.0-9.2° (12.5%), 89.2-91.2° (37.1%), residue (6.0%); judging by the d<sub>4</sub>, refraction, and ailine point, the 1st fraction consists almost entirely of paraffins, the 2nd and 3rd entirely, the 4th approx. 60%, 2,4-dimethylpentane predominates in the 1st and 2nd fraction and constitutes about 80% of the 3rd; the 1st fraction contains a small amt. of C<sub>6</sub>H<sub>6</sub>, the residue about 17% PhMe (1% of the total catalyze). By Raman line photometry, over 50% of the mixed 1st, 2nd, and 3rd fraction are 2,4-dimethylpentane, not over 10% 2-methylhexane, and not over 10% 3-methylhexane; the 4th fraction consists of about 50% unreacted II, 25% 2-methylhexane, and 25% 3-methylhexane. Ring opening evidently takes place at the 4,5-bond. (3) At 290°, under identical conditions, the ratio of the nos. of mols. of cyclopentane, methylcyclopentane, I, and II opened in 1 hr. are approx. 1.2-0.7-0.2:0.1. Thus, disubstituted cyclopentanes are opened more slowly than the monosubstituted compds., and II reacts more slowly than I owing to the presence of 4 "passive" bonds (1-2, 2-3, 3-4, 1-5) as against only 3 in I (1-2, 2-3, 1-5). IX. Hydrogenation of 1,1-dimethylcyclopentane over platinum carbon. 1/2d 185-186. (1) 1,1-Dimethylcyclopentane (III) was synthesized with a higher yield than could be obtained by methods hitherto described: 83 g. 5,5-dimethyl-1,3-cyclopentandione (IV), in 1000 ml. 93% EtOH, was hydrogenated in the presence of 17 g. Raney Ni, under an initial pressure

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**CONTINUATION**

of  $H_2$  at 75-80 atm., with gradual heating up to 180°; the pressure rose up to 140°; from 140° to 180°, it fell rapidly down to 10-20 atm. After cooling, the  $H_2$  pressure was again increased to 70-80 atm. and the autoclave was heated again to 180° for a total 6-8 hrs. after which no more  $H_2$  was absorbed. Distill. of the product gave 50-53 g. (64-68% of the theory) 1,1-dimethyl-3-cyclohexanol. This was allowed to drop into 180 ml.  $HNO_3$  (d. 1.4), 80 ml.  $H_2O$ , and 0.2 g.  $NH_4VO_3$  at 55-60°; the mixt. was then heated until cessation of evolution of N oxides, and evapd. to a syrup which on cooling gave a cryst. mass; the mixt. of  $\alpha,\alpha$ - and  $\beta,\beta$ -dimethyladipic acids (recrystd. from  $H_2O$  and dried over  $H_2SO_4$ ) was heated with 2%  $Tb(OH)_4$  to 280-300°; from 220 g. of the acids, 97.5 g. of a mixt. of 2,2- and 3,3-dimethylcyclopentanone was obtained, with a yield of 69% after redistn. at 145-8° (743 mm.). The ketones were converted into 85.1 g. hydrazones (yield, 77.6% after redistn. at 126-30° at 65 mm.). The hydrazones were decoupled by heating with fused KOH and platinized C; fractionation of the hydrocarbons gave 49.9 g. III; final yield (with respect to IV) 17.8% of theory. (2) Hydrogenation of III under the same conditions as before, with an excess of  $H_2$  (1.51  $H_2$  outgoing per hr.), at 210°, vol. rate 0.20-0.30, gave a catalyze estd. to contain 60% paraffins and fractionated, under 752.6 mm., into b. 78.7-8.25° (30.4%), 70.25-87.5° (62.1%), residue 3.7%, losses 3.7%. From the phys. consts., the 1st fraction is entirely 2,2-dimethylpentane; the 2nd fraction contains about 40% paraffins which Raman photometry shows to consist of approx. equal amts. of 2,2- (20%) and 3,3-dimethylpentane.

Thus, the total paraffins obtained consist of about 77% 2,2- and about 23% 3,3-dimethylpentane. The catalyze obtained at 217-300°, vol. rate 0.21, contained about 80% paraffins; it was fractionated (under 743.5 mm.) into b. 77.2-8.0° (37.4%), 70.0-82.8° (52.2%), residue 5.2%, losses 5.2%; the 1st fraction is, again, almost all 2,2-dimethylpentane, the 2nd fraction contains 75-80% paraffins which, by Raman photometry, consist of nearly equal amts. cf 2,2- (32%) and 3,3-dimethylpentane (38%); the total proportions of 2,2- and 3,3 isomers are thus very nearly the same (74 and 26%, resp.) at 200°. The balance of the 2nd fractions (59 and 30% at 200° and 300°, resp.) is unreacted III. (3) The rate of hydrogenation of III is very nearly the same as that of the monosubstituted methylcyclohexane. Ring opening takes place exclusively at the 2,3-, 3,4-, and 4,5-bonds, to the extent of 36-8%, 24-8%, and 36-8%, resp.; III undergoes absolutely no ring opening at the 1,2- and 1,5-bonds, in contrast to methyleclopentane where this type of ring opening does take place to some extent.

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## TABLE I. METALLURGICAL LITERATURE CLASSIFICATION

ECONOMY	SOCIETY MAP ONLY ONE	NATIONAL	EDUCATIONAL																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																				
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180	181	182	183	184	185	186	187	188	189	190	191	192	193	194	195	196	197	198	199	200	201	202	203	204	205	206	207	208	209	210	211	212	213	214	215	216	217	218	219	220	221	222	223	224	225	226	227	228	229	230	231	232	233	234	235	236	237	238	239	240	241	242	243	244	245	246	247	248	249	250	251	252	253	254	255	256	257	258	259	260	261	262	263	264	265	266	267	268	269	270	271	272	273	274	275	276	277	278	279	280	281	282	283	284	285	286	287	288	289	290	291	292	293	294	295	296	297	298	299	300	301	302	303	304	305	306	307	308	309	310	311	312	313	314	315	316	317	318	319	320	321	322	323	324	325	326	327	328	329	330	331	332	333	334	335	336	337	338	339	340	341	342	343	344	345	346	347	348	349	350	351	352	353	354	355	356	357	358	359	360	361	362	363	364	365	366	367	368	369	370	371	372	373	374	375	376	377	378	379	380	381	382	383	384	385	386	387	388	389	390	391	392	393	394	395	396	397	398	399	400	401	402	403	404	405	406	407	408	409	410	411	412	413	414	415	416	417	418	419	420	421	422	423	424	425	426	427	428	429	430	431	432	433	434	435	436	437	438	439	440	441	442	443	444	445	446	447	448	449	450	451	452	453	454	455	456	457	458	459	460	461	462	463	464	465	466	467	468	469	470	471	472	473	474	475	476	477	478	479	480	481	482	483	484	485	486	487	488	489	490	491	492	493	494	495	496	497	498	499	500	501	502	503	504	505	506	507	508	509	510	511	512	513	514	515	516	517	518	519	520	521	522	523	524	525	526	527	528	529	530	531	532	533	534	535	536	537	538	539	540	541	542	543	544	545	546	547	548	549	550	551	552	553	554	555	556	557	558	559	560	561	562	563	564	565	566	567	568	569	570	571	572	573	574	575	576	577	578	579	580	581	582	583	584	585	586	587	588	589	590	591	592	593	594	595	596	597	598	599	600	601	602	603	604	605	606	607	608	609	610	611	612	613	614	615	616	617	618	619	620	621	622	623	624	625	626	627	628	629	630	631	632	633	634	635	636	637	638	639	640	641	642	643	644	645	646	647	648	649	650	651	652	653	654	655	656	657	658	659	660	661	662	663	664	665	666	667	668	669	670	671	672	673	674	675	676	677	678	679	680	681	682	683	684	685	686	687	688	689	690	691	692	693	694	695	696	697	698	699	700	701	702	703	704	705	706	707	708	709	710	711	712	713	714	715	716	717	718	719	720	721	722	723	724	725	726	727	728	729	730	731	732	733	734	735	736	737	738	739	740	741	742	743	744	745	746	747	748	749	750	751	752	753	754	755	756	757	758	759	760	761	762	763	764	765	766	767	768	769	770	771	772	773	774	775	776	777	778	779	780	781	782	783	784	785	786	787	788	789	790	791	792	793	794	795	796	797	798	799	800	801	802	803	804	805	806	807	808	809	810	811	812	813	814	815	816	817	818	819	820	821	822	823	824	825	826	827	828	829	830	831	832	833	834	835	836	837	838	839	840	841	842	843	844	845	846	847	848	849	850	851	852	853	854	855	856	857	858	859	860	861	862	863	864	865	866	867	868	869	870	871	872	873	874	875	876	877	878	879	880	881	882	883	884	885	886	887	888	889	890	891	892	893	894	895	896	897	898	899	900	901	902	903	904	905	906	907	908	909	910	911	912	913	914	915	916	917	918	919	920	921	922	923	924	925	926	927	928	929	930	931	932	933	934	935	936	937	938	939	940	941	942	943	944	945	946	947	948	949	950	951	952	953	954	955	956	957	958	959	960	961	962	963	964	965	966	967	968	969	970	971	972	973	974	975	976	977	978	979	980	981	982	983	984	985	986	987	988	989	990	991	992	993	994	995	996	997	998	999	1000

BYLANOV, T. F.

USSR/Chemistry - Cyclopentane  
Chemistry - Hydrogenation, Catalytic

Jul/Aug 48

"Catalytic Hydrogenation of Cyclopentane-Type Hydrocarbons With Resulting Cleavage of the Cycle," B. A. Kazanskiy, T. F. Bylanov, Inst Org Chem, Acad Sci USSR, 5½ pp

"Iz Ak NaukSSSR, Otdel Khim Nauk" No 4

Cyclopentane is partially hydrogenated with ring cleavage by hydrogen evolved during the simultaneous dehydrogenation of cyclohexane, at 275° and 300° in the presence of platinized carbon. Submitted 3 Dec 1947.

FA 8/49T14

BULANOVA, T. F.

PA 35/49T11

USSR/Chemistry - Cyclopentane, Hydrogena- Sep 48  
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"The Hydrogenation of Cyclopentane With Nickel and  
Palladium Catalysts," Acad. B. A. Kazansky, T. F.  
Bulanova, 1 pp

"Dok Ak Nauk SSSR" Vol LXII, No 1

PA 35/49T11

Discusses differences in action of platinumized carbon  
and nickel as catalysts for hydrogenation, based on  
previously established data. Experiments on hydro-  
genation of cyclopentane over nickel on kieselguhr  
and over palladium, and on destructive hydrogenation

PA 35/49T11

USSR/Chemistry - Cyclopentane, Hydrogena- Sep 48  
tion of (Contd)

PA 35/49T11

of normal pentane over nickel on kieselguhr, led  
authors to conclude that platinumized carbon exerts a  
specific action in the sense that hydrogen is added  
only to two neighboring carbon atoms, whereupon  
splitting occurs without any side reactions. Sub-  
mitted 6 Jul 48.

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*Catalytic hydrogenation of cyclopentane hydrocarbons amine points. At 275°, space velocity 0.2-0.4, total with ring opening. X. Hydrogenation of cyclopentane amts. of mixt. passed 17.5 g., the mixts. I:1, 5:3, and with the hydrogen set free in simultaneous dehydrogena- 3:1 gave, resp., 2.07, 2.01, and 3.06 g. C<sub>5</sub>H<sub>10</sub>/hr., i.e. tion of cyclohexane. B. A. Kuzminskii and T. P. Butanova, a degree of conversion of I to C<sub>5</sub>H<sub>10</sub> of, resp., 34.0, 25.9, /mestr. Akad. Nauk. S.S.R., Oddel. Khim. Nauk 1948, 400-11; cf. C.A. 42, 45354. — Mixts. of cyclopentane (I) 1800, and 915 ml./hr.; the degree of dehydrogenation of and cyclohexane (III) in the molar ratios 1:3, 1:1, 3:3, II was 80-100%. Under the same conditions, pure I 3:1, and 0:1, of which 3:1 corresponds to the stoichio- with excess H<sub>2</sub> gave C<sub>5</sub>H<sub>10</sub> 4.22 g./hr., conversion 04.5%; metric ratio 3C<sub>5</sub>H<sub>10</sub> + C<sub>6</sub>H<sub>10</sub>, the 1st the 0:1 mixt., passed 3C<sub>5</sub>H<sub>10</sub> 5.40 allowed to react, without carrier gas, on platinumized C: 1:1, 5:3, and 3:1 gave, resp., 30.0, 32.0, 31.0, and 15.0 (20% Pt) at 275 and 300°. In all mixts., part of the de- conversion of II nearly 100%. Mixt. 0:1, in a stream of hydrogenation of II by the H<sub>2</sub> evolved, part of the de- C<sub>5</sub>H<sub>10</sub> 0.55 g./hr., conversion 37%. Pure I, then flushed with pure N<sub>2</sub> at the temp. of the intended hydrogenation of II; the amt. of C<sub>5</sub>H<sub>10</sub> formed per hr. reduced in H<sub>2</sub> at 150°, 15-18 ml./hr. for 5 hrs., then in the de- conversion 53.0%; after the runs with the 0:1 mixt., I gave only 5.90 hydroge- nization of II at 275°, 10-12 ml./hr. for 5 hrs., then flushed with pure N<sub>2</sub>; the catalyst suffers poisoning with respect to dehydro- nization of II although its activity with respect to dehydro- nization of II is preserved.*

N. Thor

Bulanova, I. F.

Grad Chem Sci

Dissertation: "Catalytic Hydrogenation of Cyclopentene."

9 June 49

Inst of Organic Chemistry, Acad Sci USSR

SO Vecheryaya Moskva  
Sum 71

Bulanova, T. F.

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Optical investigation of hydrocarbons. V. Raman spectra of some napthenes  
and nonanes.

BULANOVA, T. F.

Determination of individual hydrocarbon composition of  
gasolines by the combined method. II. Two gasolines  
from petroleum of Kazanbulak origin. B. A. Kazanskiy, A. F. Plat.,  
Ye. A. Mikhaylova. A. L. Liberman, M. I.  
Batuyev, T. F. Bulanova, and G. A. Tarasova, (N. D. Zelinskiy  
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266-77; cf. C.A. 45, 7342a. -- Two specimens of gasoline from  
Kazanbulak area were examined by the combined  
optical-distn. method. Fractions b. under 150° over 70  
hydrocarbons were identified, thus accounting for 40-50%  
of the total compn. It is shown that despite the close  
origin of the specimens geographically, considerable differences in compn. are found. III. Surakhan gasolines.  
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man, Ye. A. Mikhaylova, F. A. Bazhulin, M. I. Batuyev,  
S. A. Ukholin, T. F. Bulanova, and G. A. Tarasova. Ibid.  
278-91. -- Two specimens of Surakhan gasolines were examined by the  
combined method. In both some 47 hydrocarbons were identified,  
accounting for 77-84% of the total compn. Distn. curves and  
distn., data are cited.

G. M. Kosolapoff

BULANOVA, T. F.

U S S R .

Determination of individual hydrocarbons in gasolines by the combined method. VI. Karachukhur gasoline. B. A. Kazanskiy, G. S. Landsberg, A. F. Flate, A. L. Liberman, Ye. A. Mikhaylova, Kh. Ye. Steyn, T. F. Bulanova, G. A. Tarasova, and V. T. Aleksayan (N. D. Zelinskiy Inst., Org. Chem. Acad. Sci. U.S.S.R., Moscow). Izvest. Akad. Nauk S.S.R., Otdel. Khim. Nauk, 1954, 1053-66; cf. C. A. 45, 7342b. ---The combination of distn. chromatography, and Raman spectroscopy applied to a sample of Karachukhur gasoline (150° end point) was successful in identifying 8514% of the hydrocarbon compn., showing the presence of 63 hydrocarbons. The gasoline contained 16.37% aromatic, and approx. equal amts. of aliphatic and naphthenic hydrocarbons; about 40% of the paraffins are normal alkanes. The ratio of cyclopentane derivs. to cyclohexane derivs. is 0.4%. G. M. Kosolapoff

BULANOVA, T. F.

USSR/ Physics - Spectral analysis

Card 1/1 Pub. 43 - 36/62

Authors : Kazanskiy, B. A.; Landsberg, G. S.; Aleksanyan, V. T.; Bulanova, T. F.; Liberman, A. L.; Mikhaylova, Ye. A.; Plate, A. F.; Sterin, Kh. Ye.; and Ukholin, S. A.

Title : Analysis of aromatic ligroin parts by the combined diffusion spectra

Periodical : Izv. AN SSSR. Ser. fiz. 18/6, 704-706, Nov-Dec 1954

Abstract : Brief report is presented on the method and some results obtained during individual and close-group analysis of primary and secondary aromatics of ligroin. Analysis of results obtained showed that the basic ligroin (taken from the Embensk Petroleum Source) contained alkyl substitutes of benzene and cyclohexane with short term substituting radicals. Three references: 1 USA and 2 USSR (1947-1953). Tables.

Institution : Acad. of Sc., USSR., The N. D. Zelinskiy Inst. of Organ. Chem. and the Commission on Spectroscopy

Submitted : .....

BULANOVA, T. F.

Catalytic transformations of *n*-heptane and *n*-octane in the presence of platinised charcoal. B. A. Kuznetsov, V. I. Liberman, T. F. Bulanova, V. T. Alaksanyan and Kh. E. Stepin (Dokl. Akad. Nauk SSSR) 1951, No. 77-80. *n*-Heptane and *n*-octane are passed at 300° at different rates over platinised charcoal containing 20% Pt or 20% Pt and 2% Fe, and the products of the reaction are analysed. The products contain 10-11% of alicyclic, + 2% of unsaturated, and 1-5% of aromatic hydrocarbons. A certain amount of aliphatic branched-chain isomers is also formed but the bulk of the hydrocarbons passes unchanged through the catalyst.  
S. K. Lachowicz.

LANDSBERG, Grigoriy Samuilovich, akademik [deceased]; KAZANSKIY, Boris Aleksandrovich, akademik; BAZHULIN, P.A., doktor fiziko-matemat. nauk; BULANOVA, T.F.; LIBERMAN, A.L., MIKHAYLOVA, Ye.A.; PLATE, A.F.; STERIN, Kh.Ue.; SUSHCHINSKIY, M.M.; TARASOVA, G.A.; UKHOLIN, S.A.; BRUSOV, I.I., red.izd-va; KASHINA, P.S., tekhn.red.

[Determination of the individual hydrocarbon composition of straight-run gasolines by the combined method] Opredelenie individual'nogo uglevodorodnogo sostava benzinov priamoi gonki kombinirovannym metodom. Moskva, Izd-vo Akad.nauk SSSR, 1959.  
(MIRA 12:8)  
362 p.

(Gasoline)

S/020/62/147/005/021/032  
B106/B186

AUTHORS: Eydus, Ya. T., Bulanova, T. F., Sergeyeva, N. S.

TITLE: Zirconium and titanium dioxides - promoters of the cobalt catalyst in the synthesis of higher hydrocarbons from carbon monoxide and hydrogen

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 147, no. 5, 1962, 1105 - 1107

TEXT: The activating effect of  $ZrO_2$  and  $TiO_2$  on a cobalt-kieselguhr (1:1) catalyst was studied for the synthesis of higher hydrocarbons from CO and  $H_2$ . All experiments were made in a continuous flow system at atmospheric pressure and various temperatures (20 - 30 hrs reaction time at each temperature). The initial gas mixture contained CO and  $H_2$  at a ratio of 1:2. Fig. 1 shows the results obtained. It has been found that catalysts containing 18%  $TiO_2$  or  $ZrO_2$  are more active than the known catalyst with 18%  $ThO_2$ . At optimum reaction temperatures (195°C & 2100°C) of the catalyst activated with 18%  $TiO_2$ , not merely low hydrocarbons of the type  $C_2-C_4$  are

Card 1/2

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p. 1

Zirconium and titanium...

S/020/62/147/005/021/032  
B106/B186

formed but higher hydrocarbons (from C<sub>5</sub> upward) at the volume ratio 0.6:1 (gasoline : oil) as compared with 18 : 1 at the optimum reaction temperature (210°C) of the non-activated catalyst, and 0.9:1 gasoline - oil ratios obtained with the catalyst activated by 18% ThO<sub>2</sub>. There are 1 figure and 1 table. The English-language reference is: S. Kodama, Sci. pap. Inst. Phys. Chem. Res., 14, 253 (1930).

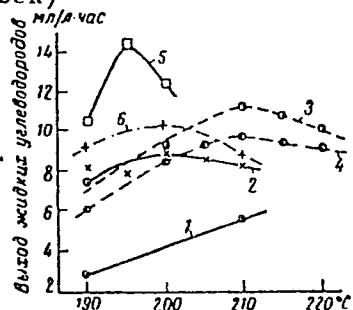
ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo Akademii nauk SSSR (Institute of Organic Chemistry imeni N. D. Zelinskogo of the Academy of Sciences USSR)

PRESENTED: August 3, 1962, by B. A. Kazanskiy,  
Academician

SUBMITTED: July 12, 1962

Fig. 1. Activation of Co-kieselguhr (1:1) catalyst.  
Legend: Ordinate: liquid hydrocarbon yield,  
ml/l·hr; (1) catalyst without promoter; (2) 12%  
ZrO<sub>2</sub>; (3) 18% ZrO<sub>2</sub>; (4) 24% ZrO<sub>2</sub>; (5) 18% TiO<sub>2</sub>;  
(6) 18% ThO<sub>2</sub>. (for comparison).

Card 2/2



BULANOVA, T.F.; EYDUS, Ya.T.; SERGEYEVA, N.S.; KHUDYAKOV, Yu.T.

Directed catalytic synthesis of solid paraffins from carbon monoxide and hydrogen. Dokl. AN SSSR 153 no.1:101-103 N '63.  
(MIRA 17:1)

1. Institut organicheskoy khimii im. N.D. Zelinskogo AN  
SSSR. Predstavлено академиком B.A. Kazanskim.

ACCESSION NR: AP4024404

S/0204/64/004/001/0061/0067

AUTHOR: Eydus, Ya. T.; Bulanova, T. F.; Sergeyeva, N. S.

TITLE: The promoting effect of zirconium dioxide on the cobalt catalyst in the synthesis of higher hydrocarbons from carbon monoxide and hydrogen at atmospheric pressure.

SOURCE: Neftekhimiya, v. 4, no. 1, 1964, 61-67

TOPIC TAGS: hydrocarbon synthesis, oxo synthesis, cobalt catalyst, zirconium dioxide, thorium dioxide, promoter, gasoline synthesis, hydrocarbon oil synthesis

ABSTRACT: The promoting effect of  $ZrO_2$  on the cobalt catalyst on a kieselguhr carrier in the synthesis of higher hydrocarbons from  $CO$  and  $H_2$  at atmospheric pressure was investigated in view of its similarity to  $ThO_2$ , a known promoter. Of the preliminary catalysts investigated (Co-kieselguhr, Co-kieselguhr with  $ZrO_2$  and Co-MgO-kieselguhr with  $ZrO_2$ ), the latter, containing MgO proved most active. At 200-240 C after 300 hours operation without regeneration it still yielded 70 gm/m<sup>3</sup> of hydrocarbons. Catalysts with various component ratios were examined; the most active found was 100Co:6 $ZrO_2$ :10MgO:200 kieselguhr. The yield varied

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ACCESSION NR: AP4024404

depending on the length of time the catalyst was used, e.g., after the first 15 hours after catalyst preparation, at 190 C for 170 hours, the yield of higher hydrocarbons was 111-125 gm/m<sup>3</sup> with gasoline/oil ratio of 0.55; during the next 400 hours, the yield was 93.3 with 0.9 ratio. A similar catalyst prepared with ThO<sub>2</sub> promotor instead of ZrO<sub>2</sub> yielded, after the first 20 hours, 105 gm/m<sup>3</sup> of product with gasoline/oil ratio of 0.45. Orig. art. has: 4 tables and 1 figure.

ASSOCIATION: Institut organicheskoy khimii AN SSSR im. N. D. Zelinskogo  
(Institute of Organic Chemistry, AN SSSR)

SUBMITTED: 08Jan63

DATE ACQ: 17Apr64

ENCL: 00

SUB CODE: GC

NO REF SOV: 004

OTHER: 006

Card 2/2

ROZENGART, M.I.; BULANOVA, T.F.

Problem of the regeneration of alumina-chromia catalysts for  
dehydrocyclization. Zhur. prikl. khim. 37 no.2:383-388 F '64.  
(MIRA 17:9)

L 36484-65 EPP(c)/ENT(n)/T Pr-4 RM

UR/0204/64/004/005/0763/0766

25  
B

ACCESSION NR: AP5010563

AUTHOR: Bulanova, T. F.; Eydus, Ya. T.; Sergeyeva, N. S.

TITLE: Promoting action of titanium dioxide on a cobalt catalyst in the reaction of synthesis of higher hydrocarbons at atmospheric pressure from carbon monoxide and hydrogen

SOURCE: Neftekhimiya, v. 4, no. 5, 1964, 763-766

TOPIC TAGS: catalysis, titanium, inorganic oxide, cobalt, hydrocarbon, organic synthetic process

Abstract: Titanium dioxide, like thorium dioxide and zirconium dioxide, was found to have a promoting action on Co-kieselguhr and Co-MgO-kieselguhr catalysts in the reaction of formation of liquid hydrocarbons from carbon monoxide and hydrogen at atmospheric pressure. Catalysts simultaneously containing TiO<sub>2</sub> and MgO were more active than a catalyst containing only TiO<sub>2</sub>. Just as in the case of promotion by ZrO<sub>2</sub>, a catalyst containing 100 Co:6 TiO<sub>2</sub>: 10 MgO:200 kieselguhr proved to be optimum. Orig. art. has 2 tables.

ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo AN SSSR (Institute of Organic Chemistry, AN SSSR)

SUBMITTED: 07Oct63

ENCL: 00

SUB CODE: OC, GC

NO REF Sov: 005

OTHER: 001

JPRS

Card 1/1

L 33999-65 EWT(m)/EPF(c)/EWP(j)/T PC-4/Pr-4 RM  
ACCESSION NR: AP5005079 S/0204/65/005/001/0068/0075

AUTHOR: Eydus, Ya. T.; Bulanova, T. F.; Muzovskaya, O. A.; Sergeyeva, N. S. 29  
36

TITLE: Catalytic synthesis of high-molecular hydrocarbons from carbon monoxide and hydrogen in the presence of Co-MgO-kieselguhr catalysts, activated with zirconium or titanium dioxide

SOURCE: Neftekhimiya, v. 5, no. 1, 1965, 68-75

TOPIC TAGS: hydrocarbon synthesis, catalytic hydrogenation, carbon monoxide, hydrogen exchange, cobalt catalyst, magnesium oxide, kieselguhr, zirconium dioxide, titanium dioxide, paraffin synthesis

ABSTRACT: The authors studied the formation of solid paraffins by Fischer-Tropsch synthesis on zirconium- or titanium dioxide activated cobalt-magnesium oxide-kieselguhr catalysts. Catalysts having the composition 200 parts kieselguhr/100 parts Co/6-10 parts ZrO<sub>2</sub> or TiO<sub>2</sub>/6-10 parts MgO were obtained by precipitation of nitrates on the kieselguhr support, reduced at 400°C and 1 atm. H<sub>2</sub> pressure, and used as catalysts at 10 atm., 100 hr<sup>-1</sup> flow rate and a 1:2 ratio of carbon monoxide to H<sub>2</sub>, as well as in non-continuous tests and at atmospheric pressure. Synthesis

Cord 1/2

L 33999-65

ACCESSION NR: AP5006079

at atmospheric pressure gave primarily liquid hydrocarbons, as did synthesis on thorium-activated catalysts, while synthesis at 10 atm. gave, after a development period of 3-8 days, 100-110 g/m<sup>3</sup> of solid paraffin waxes which contained 20-30% liquid and 70-75% solid hydrocarbons; 15-20% of the solid fraction had melting points of 106-116°C. Liquid and solid reaction products were fractionated and the physical and chemical characteristics of individual fractions are given. Orig. art. has: 5 tables, 1 figure and 1 formula.

3

ASSOCIATION: Institut organicheskoy khimii im. N. D. Zelinskogo, AN SSSR (Organic chemistry institute, AN SSSR); Tsentral'naya laboratoriya Redkinskogo optychnogo zavoda (Central laboratory, Redkinsk experimental plant); Komiteata khimicheskoy promyshlennosti pri Gosplane SSSR (Chemical industry committee, State planning commission, SSSR)

SUBMITTED: 28Jan64

ENCL: 00

SUB CODE: OC

NO REF SOV: 012

OTHER: 009

Card 2/2

BULANOVA, V.A.; KHOLODILOV, V.V.

Some examples of the interpretation of cementograms. Neft.  
khoz. 43 no. 2-26-30 F '65. (MIRA 18:4)

BULANOVA, V.A.; KHOLODILOV, V.V.

Using caliper logging in the solution of certain geological  
problems. Geol. nefti i gaza 8 no.11:35-37 N '64.

(MIRA 17:12)

1. Volgogradneftegeofizika.

L 21825-65 EPA/EPP(c)/EPP(n)-2/EPR/EPA(s)-2/EWT(m)/EWP(b)/EWP(t) Pr-4/Pt-10/  
pu-4/Paa-4 SSD(a)/IJP(c) WJW/JWD/JD  
ACCESSION NR: AP5001756 S/0153/64/007/005/0862/0863

AUTHOR: Shidlovskiy, A. A.; Shmagin, L. F.; Bulanova, V. V.

TITLE: Burning of ammonium perchlorate under atmospheric pressure

SOURCE: IVUZ. Khimiya i khimicheskaya tekhnologiya, v. 7, no. 5, 1964,  
862-863

TOPIC TAGS: ammonium perchlorate, catalyst, ammonium perchlorate de-  
composition, ammonium perchlorate burning

ABSTRACT: The catalytic effect of  $\text{Cu}_2\text{O}$ ,  $\text{Cu}_2\text{Cl}_2$ ,  $\text{CuO}$ ,  $\text{CuCO}_3$ ,  $\text{MnO}_2$ ,  $\text{MnCO}_3$ ,  $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$ ,  $\text{Co}_2\text{O}_3$ ,  $\text{ZnO}$ ,  $\text{Fe}_2\text{O}_3$ ,  $\text{NiO}$ ,  $\text{Ni}_2\text{O}_3$ ,  $\text{Cr}_2\text{O}_3$ ,  $\text{Cu}$ ,  $\text{Cr}_2\text{O}_4$ ,  $\text{CdO}$ , or  $\text{MgO}$  on the thermal decomposition and burning of ammonium perchlorate has been studied at atmospheric pressure. The experiments were conducted with technical-grade  $\text{NH}_4\text{ClO}_4$  sifted through a no. 61 sieve and containing 5% of the ground pure catalysts. The mixtures were burned at 20 and 100°C in glass tubes. At 20°C,  $\text{NH}_4\text{ClO}_4$  burns in the presence of  $\text{Cu}_2\text{O}$ ,  $\text{CuO}$ ,  $\text{Cu}_2\text{Cl}_2$ ,  $\text{MnO}_2$ , or  $\text{MnCO}_3$ , and at 100°C

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ACCESSION NR: AP5001756

in the presence of CuCO<sub>3</sub>, MnCl<sub>2</sub>·4H<sub>2</sub>O, Co<sub>2</sub>O<sub>3</sub>, or ZnO. The highest burning velocity and highest thermal coefficient of the burning velocity (0.60—0.80 mm/sec at 20°C and 1.40—2.08 mm/sec at 100°C) are exhibited by mixtures containing copper compounds. Mixtures with Fe<sub>2</sub>O<sub>3</sub>, NiO, Ni<sub>2</sub>O<sub>3</sub>, Cr<sub>2</sub>O<sub>3</sub>, Cu, Cr<sub>2</sub>O<sub>4</sub>, CdO as MgO do not burn under the above conditions. Orig. art. has: 1 table.

ASSOCIATION: Moskovskiy institut khimicheskogo mashinostroyeniya  
(Moscow Institute of Chemical Machinery)

SUBMITTED: 03Apr64

ENCL: 00

SUB CODE: GC, FP

NO REF SOV: 009

OTHER: 004

ATD PRESS: 3166

Card 2/2